A Novel Conversion of 2,4-Diaryl-2,3-dihydro-1H-1,5-benzodiazepines into 2,4-Diaryl-3H-1,5-benzodiazepines

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A novel conversion of 2,4-diaryl-2,3-dihydro-1H-1,5-benzo-diazepins into 2,4-diaryl-3H-1,5-benzodiazepines by the reaction with m-chloroperbenzoic acid (MCPBA) was reported.

Keywords 1H-1,5-Benzodiazepine, 3H-1,5-benzodiazepine, m-chloroperbenzoic acid, conversion

Introduction

In recent three decades many new pharmaceuticals synthesized have structures containing heterocyclic rings, especially benzodiazepine. Benzodiazepine derivatives are one of the most important classes of bioavailible therapeutic agents having widespread biological activities including anxiolytic, anticonvulsant, and antihypnotic activities. 1 They also act as selective cholecystokinin (CCK) receptor subtype A and B antagonists, 2 plateletactivating factor antagonists, human immunodeficiency virus trans-activator Tat/Tar antagonists,3 and as farnesyltransferase inhibitors. 4 During recent years, our research group has focused on studies of the synthesis and stereo-structure of novel 1,5-benzothiazepine and 1,5benzodiazepine derivatives in order to develop new pharmaceuticals. 5-9 The present work deals with the conversion of 2, 4-diaryl-2, 3-dihydro-1H-1, 5-benzodiazepins into 2,4-diaryl-3H-1,5-benzodiazepines by reaction with m-chloroperbenzoic acid (MCPBA).

Results and Discussion

2, 4-Disubstituted 3H-1, 5-benzodiazepine deriva-

tives, such as clozapine, chlorpromazine, etc., showed potential neuroleptic activity. ¹⁰ Synthesis of 2, 4-disubstituted 3H-1,5-benzodiazepines has been previously reported. ¹¹⁻¹⁴ The two most common methods are Micheal additions of o-diaminobenzene derivatives with α -carbonyl alkynes followed by intramolecular condensation reactions of amines and ketones ^{11,12} or condensation of o-diaminobenzene derivatives and 1,3-diketones. ^{13,14}

We herein report a novel method to convert 2,4-diaryl-2,3-dihydro-1H-1,5-benzodiazepins into 2,4-diaryl-3H-1,5-benzodiazepines by oxidation of m-chloroperbenzoic acid and dehydrolysis in a one-pot reaction.

m-Chloroperoxybenzoic acid (MCPBA) is a useful peroxy-acid. It has been widely used to synthesize epoxy compounds and oxaziridine derivatives by oxidation of olefins and imines, 15,16 respectively. However, the use of a buffered aqueous medium seems to be especially suitable for oxidation of somewhat acid-sensitive starting materials or products. 17 We carried out our experiments by a basic biphasic oxidation procedure in a mixture of dichloromethane and saturated aqueous sodium bicarbonate by applying a phase transfer catalyst benzyltriethylammonium chloride (TEBA) because the imine group in 1,5-benzodiazepine is acid-sensitive. 2,4-Diaryl-2,3dihydro-1H-1, 5-benzodiazepines 1 can react readily with MCPBA to yield 2, 4-diaryl-3H-1, 5-benzodiazepines 2 in moderate yields. This is a new method for synthesis of 3H-1, 5-benzodiazepine derivatives and is also a novel conversion of a secondary amine into an imine derivative.

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Table 1 Conversion of 2,4-diaryl-2,3-dihydro-1*H*-1,5-benzodiazepines 1 into 2,4-diaryl-3*H*-1,5-benzodiazepines 2

Entry	R ¹	R ²	Yield of 2 (%)	m.p. of 2 (℃)/(Lit)
а	Ph	Ph	63	140—141(139—140 ¹²)
b	3-ClPh	Ph	68	$170-171(171^{13})$
c	4-ClPh	Ph	72	176—177(176—177 ¹²)
d	2-MeOPh	Ph	54	$102-103^a(102-103^{18})$
e	4-MeOPh	Ph	58	153—154(153—154 ¹²)
f	2-BrPh	Ph	45	172-174
g	3-BrPh	Ph	60	154—155 (154 ¹³)
h	Ph	4-MeOPh	52	153—154(153—154 ¹²)

^aIn the same reaction conditions, **1d** was converted into 2,3-dihydo-2-hydroxy-2-(2-methoxylphenyl)-4-phenyl-1*H*-1,5-benzodiazepine **3d**, which was further refluxed in benzene in the presence of *m*-chlorobenzoic acid to yield **2d** in 54% combined yield. Aminohydrins are a kind of unstable compounds. Compound **3d** is a stable aminohydrin probably due to its intramolecular hydrogen-bonding.

In order to extend the application of this reaction, we also carried out the oxidation of several secondary amines, such as $PhCH_2NHPh$, $PhCH_2NHPhOMe-p$, $(PhCH_2)_2NH$, etc. However, no expected products have been isolated in these reaction conditions. For $PhCH_2NHPh$ and $PhCH_2NHPhOMe-p$, only nitrone derivatives $PhCH = N(\rightarrow O)$ Ph and $PhCH = N(\rightarrow O)$ PhOMe-p were obtained. ¹⁹ But $(PhCH_2)_2NH$ was converted into hydroxyamine derivative, $(PhCH_2)_2NOH$.

All products were characterized by 1H NMR spectra and the melting points compared with those in the literatures. 12,13,18,19 The unknown products 2f and 3d were also characterized by IR and MS spectrometries and elemental analysis. In summary, a novel conversion of 2,4-diaryl-2,3-dihydro-1H-1,5-benzodiazepines into 2,4-diaryl-3H-1,5-benzodiazepines by oxidation of m-chloroperbenzoic acid (MCPBA) in moderate yields was reported.

Experimental

Melting points were obtained on a Yanaco melting point apparatus and are uncorrected. Elemental analyses were carried out on an Elementar Vario EL elemental analyzer. The ¹H NMR spectra were recorded on a Varian Mercury 200 spectrometer in CDCl₃ with TMS as the internal standard. The IR spectra were taken on a Brucker Vector 22 FT-IR spectrophotometer in KBr. Mass spectra were obtained on a VG ZAB-HS mass spectrometer. TLC separations were performed on silica gel G plates with petroleum ether $(30-60\,^{\circ}\text{C})$ /ethyl acetate (5:1), and the plates were visualized with UV light and/or iodine vapor.

2, 4-Diaryl-2, 3-dihydro-1H-1, 5-benzodiazepines were synthesized according to the literature method. $^{5-7,20}$

Conversion of 2, 4-diaryl-2, 3-dihydro-1H-1, 5-benzodiazepins into 2, 4-diaryl-3H-1, 5-benzodiazepines

In a 50 mL three-necked flask equipped with a magnetic stirrer and a dropping funnel were placed 1.0 mmol of the aqpropriate 2,4-diaryl-2,3-dihydro-1*H*-1,5-benzodiazepine 1 in 7 mL of CH₂Cl₂, 20 mL of saturated aqueous NaHCO₃ and 0.05 g (0.25 mmol) of TE-BA (benzyltriethylammonium chloride). The solution was cooled to 0—5°C with an ice bath and 2.0—3.0 mmol of MCPBA (*m*-chloroperbenzoic acid) in 9 mL of CH₂Cl₂ was added dropwise with rapid stirring over 1 h. After the addition was complete, the solution was stirred

for additional 4—10 h at room temperature, and the CH₂Cl₂ solution was washed subsequently with water (25 mL), 10% Na₂SO₃(3×25 mL), 10% NaHCO₃(3×25 mL) and water (25 mL). After the solution was dried over anhydrous K_2 CO₃, the solvent was removed on the rotatory evaporator to give a brown residue. After crystallizing from a mixture of benzene and methanol or separating on a silica gel column with petroleum ether (30—60°C) /ethyl acetate (5:1) as an eluent, the respective products were obtained.

In the same reaction conditions, 1d was converted into 2, 3-dihydo-2-hydroxy-2-(2-methoxylphenyl)-4-phenyl-1*H*-1,5-benzodiazepine 3d in 60% yield, which was further refluxed in benzene in the presence of *m*-chlorobenzoic acid to give 2d in 90% yield.

2-(2-Bromophenyl)-4-phenyl-3H-1, 5-benzodiaze-pine (2f) Yield: 45%; m. p. 172—174°C; 1 H NMR (CDCl₃, 200 MHz) δ : 3.27 (s, br, 2H, CH₂), 6.02—7.98 (m, 13H, ArH); IR (KBr) ν : 1595 cm⁻¹; EI-MS (70 eV) m/z: 374 (M⁺), 376 (M + 2)⁺; Anal. calcd for C₂₁H₁₅N₂Br: C 67.21, H 4.03, N 7.46; found C 66.99, H 4.29, N 7.24.

2,3-Dihydo-2-hydroxy-2-(2-methoxylphenyl)-4-phenyl-1H-1,5-benzodiazepine (3d) Yield: 60%; m.p. 134—135%; 1 H NMR (CDCl₃, 200 MHz) δ : 3.47 (s, 3H, MeO), 3.96 (s, 2H, CH₂), 6.04 (s, 1H, NH), 6.32—7.97 (m, 13H, Aromatic), 12.73 (s, 1H, OH, forming H-bond with ArOMe); IR (KBr) ν : 3450, 3370, 3061, 1633, 1595 cm⁻¹; EI-MS (70 eV) m/z: 344 (M⁺), 326 (M⁺-18); Anal. calcd for $C_{22}H_{20}N_2O_2$: C 76.24, H 4.57, N 8.47; found C 76.11, H 4.87, N 8.20.

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